



Tiffany Jones

@apbiopenguins





AP Biology students are penguins because they are Dressed for Success!
You are now an AP Bio Penguin!

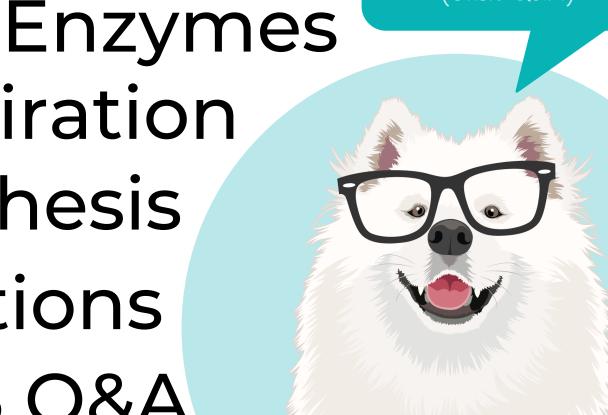




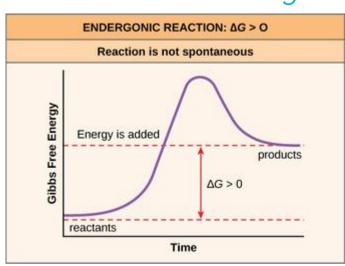
Today's Plan:

Cellular Respiration
Photosynthesis
Practice Questions
Unit 3 Q&A

Special Thank You to Mrs. McClinton (Chat Q&A)



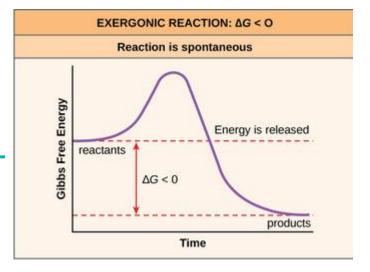
Endergonic Reaction



Change in Gibbs

Free Energy

- Not spontaneous
- ABSORB energy
- Example: $ADP + P_i \rightarrow ATP$
 - Spontaneous
 - RELEASE energy
 - Example: ATP \rightarrow ADP + P_i



Exergonic Reaction

Gibbs Free Energy & Reactions

Temperature (K)

Change in

Enthalpy

 $\Delta G = \Delta H - T\Delta S$ Change in Entropy

What is Gibbs Free Energy?

• Energy available to do work

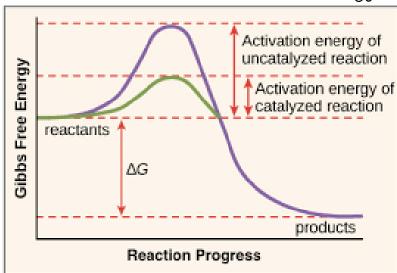
$$\Delta G = \Delta G_f - \Delta G_i$$



Enzymes

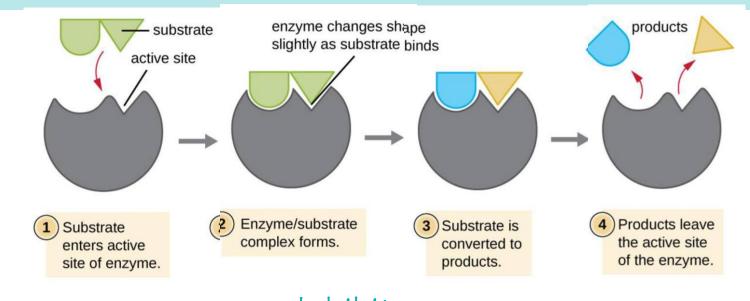
Functions

- Biological catalyst
- Speeds up chemical reactions
- Reduces the activation energy



Important Notes:

- Enzymes are PROTEINS
- Are NOT consumed by the reaction
- Have no effect on the change in Gibbs Free Energy



Competitive:

• Binds to active site

Noncompetitive:

Binds to allosteric site

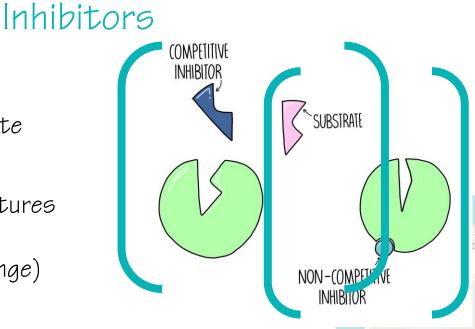
Denaturation

Environmental Temperatures

pH

(outside of optimal range)

Salinity



Cellular Respiration

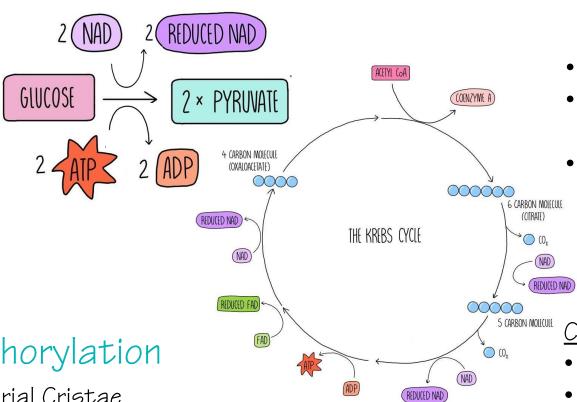
Glycolysis

- Location: Cytosol
- Starting Material:
 Glucose
- Products:

2 Pyruvate

2 NADH

2 ATP



Krebs Cycle

- Location: Mitochondrial Matrix
- Staring Material:

Acetyl CoA

Products:

 $2 CO_{2}$

3 NADH

1 FADH₂

1 ATP

Oxidative Phosphorylation

- Location: Mitochondrial Cristae
- Starting Material:

NADH/FADH₂ (electrons)

• Product:

ATPS

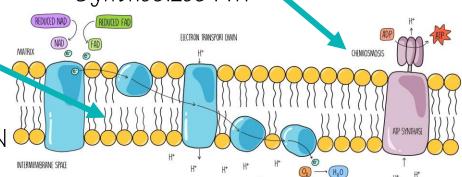
• Two Parts:

Electron Transport Chain

- Protons pumped into IM space
- Generates proton gradient
- Final electron acceptor: OXYGEN

Chemiosmosis

- ATP Synthase uses proton gradient
- Synthesizes ATP



Electron Transport Chain & Chemiosmosis

Cellular respiration includes the metabolic pathways of glycolysis, the Krebs cycle, and the electron transport chain, as represented in the figures. In cellular respiration, carbohydrates and other metabolites are oxidized, and the resulting energy-transfer reactions support the synthesis of ATP.

(a) Using the information above, describe ONE contribution of each of the following in ATP synthesis.

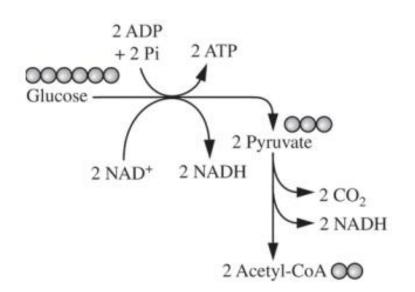


Figure 1. Glycolysis and pyruvate oxidation

Catabolism of glucose in glycolysis and pyruvate oxidation

- Produces NADH for use in ETC
- Produces acetyl-CoA for entry into Krebs cycle
- Provides energy for (substrate level) phosphorylation of ADP

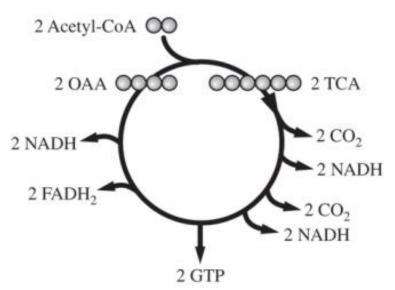


Figure 2. Krebs cycle

Oxidation of intermediates in the Krebs cycle

- Produces NADH or FADH2 for use in ETC
- Releases high energy electrons for use in ETC
- Provides energy to pump protons against their concentration gradient
- Produces GTP for (substrate level) phosphorylation of ADP

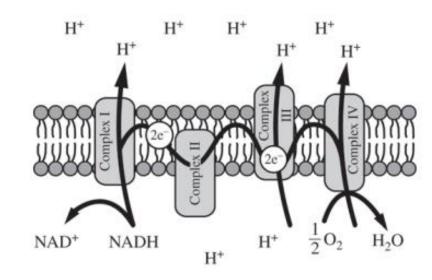


Figure 3. Electron transport chain

Formation of a proton gradient by the electron transport chain

- The flow of protons through membrane-bound ATP synthase generates ATP
- Provides energy for (oxidative) phosphorylation of ADP

Photosynthesis

Light Reactions

- Location: Thylakoid Membrane
- Starting Material:
 Water (electrons)

Products:

ATP

Photons (energy)

- Linear Electron Flow
- PSI&PSII
- Synthesizes ATP & NADPH
 - Cyclic Electron Flow
 - PS I ONLY
 Synthesizes ATP ONLY
- NADPH Synthesizes ATP ONLY

Electron Transport Chain

Protons are pumped into the thylakoid space

Calvin Cycle

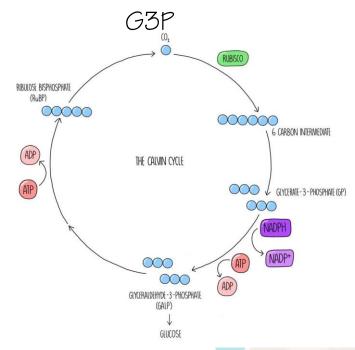
- Location: Stroma
- Starting Material:

3 CO₂

9 ATP

6 NADPH

• Products:





Multiple Choice Practice

The chemical reaction for photosynthesis is

$$6 CO_2 + 12 H_2O + light energy \rightarrow C_6H_{12}O_6 + 6 O_2 + 6 H_2O$$

If the input water is labeled with a radioactive isotope of oxygen, ¹⁸O, then the oxygen gas released as the reaction proceeds is also labeled with ¹⁸O. Which of the following is the most likely explanation?

- a. During the light reactions of photosynthesis, water is split, the hydrogen atoms combine with the CO_2 , and oxygen gas is released.
- b. During the light reactions of photosynthesis, water is split, removing electrons and protons, and oxygen gas is released.
 - c. During the Calvin cycle, water is split, regenerating NADPH from NADP+, and oxygen gas is released.
 - d. During the Calvin cycle, water is split, the hydrogen atoms are added to intermediates of sugar synthesis, and oxygen gas is released.



Organism	Temperature (°C)	Average respiration (mL O ₂ /g/min)
Mouse	10	0.0518
Mouse	25	0.0321
Cricket	10	0.0013
Cricket	25	0.0038

An experiment to measure the rate of respiration in crickets and mice at 10° C and 25° C was performed using a respirometer, an apparatus that measures changes in gas volume. Respiration was measured in mL of O_2 consumed per gram of organism over several five-minute trials and the following data were obtained.

According to the data, the mice at 10°C demonstrated greater oxygen consumption per gram of tissue than did the mice at 25°C. This is most likely explained by which of the following statements?

- a. The mice at 10°C had a higher rate of ATP production than the mice at 25°
- b. The mice at 10°C had a lower metabolic rate than the mice at 25°C.
- c. The mice at 25°C weighed less than the mice at 10°C.
- d. The mice at 25°C were more active than the mice at 10°C.



Researchers hypothesize that the plant compound resveratrol improves mitochondrial function. To test this hypothesis, researchers dissolve resveratrol in dimethyl sulfoxide (DMSO). The solution readily passes through cell membranes. They add resveratrol solution to mammalian muscle cells growing in a nutrient-rich solution (culture medium) that contains glucose. They measure ATP production at several time points after the addition of the resveratrol solution and find an increase in ATP production by the muscle cells.

- (a) **Describe** the primary advantage for a mammalian muscle cell in using aerobic respiration over fermentation.
- More ATP (per glucose molecule) is produced by aerobic respiration.





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(b) **Identify** an appropriate negative control for this experiment that would allow the researchers to conclude that ATP is produced in response to the resveratrol treatment.

Accept one of the following:

- The researchers must run the experiment without adding resveratrol.
- The researchers must treat the cells with DMSO alone.



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(c) **Predict** the effect on short-term ATP production when resveratrol-treated mammalian muscle cells are grown in a culture medium that lacks glucose or other sugars.

Accept one of the following:

- No ATP production
- Reduced ATP production



Researchers hypothesize that the plant compound resveratrol improves mitochondrial function. To test this hypothesis, researchers dissolve resveratrol in dimethyl sulfoxide (DMSO). The solution readily passes through cell membranes. They add resveratrol solution to mammalian muscle cells growing in a nutrient-rich solution (culture medium) that contains glucose. They measure ATP production at several time points after the addition of the resveratrol solution and find an increase in ATP production by the muscle cells.

- (d) The researchers find that resveratrol stimulates the production of components of the electron transport chain. The researchers claim that treatment with resveratrol will also increase oxygen consumption by the cells if glucose is not limiting. **Justify** this claim.
 - More electrons can be transferred so that more oxygen is required as the final electron acceptor.





Q & A



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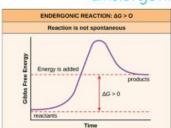




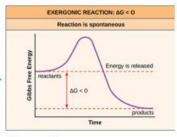
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Endergonic Reaction

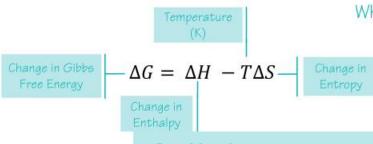


- Not spontaneous
- · ABSORB energy
- Example: ADP + P, → ATP
 - Spontaneous
 - RELEASE energy
 - Example: ATP → ADP + P₁



Exergonic Reaction

Gibbs Free Energy & Reactions



What is Gibbs Free Energy?

· Energy available to do work

 $\Delta G = \Delta G_f - \Delta G_i$



Cellular Respiration

Glycolysis

· Location: Cytosol

 Starting Material: Glucose

· Products:

2 Pyruvate

2 NADH 2 ATP

· Starting Material:

ATPs

Product:

(REDUCED NAD) GLUCOSE 2 × PYRUVATE THE KREBS CYCLE Oxidative Phosphorylation Location: Mitochondrial Cristae

Electron Transport Chain

Protons pumped into IM space

Final electron acceptor: OXYGEN

Generates proton gradient

Krebs Cycle

· Location: Mitochondrial Matrix

· Staring Material:

Acetyl CoA

Products:

2 CO2 3 NADH

1 FADH₂

1 ATP

Ch<u>emiosmosis</u>

· ATP Synthase uses proton gradient

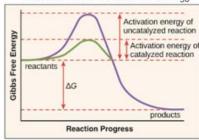
Synthesizes ATP

Enzymes

active site

Functions

- Biological catalyst
- Speeds up chemical reactions
- Reduces the activation energy



Important Notes:

- · Enzymes are PROTEINS
- · Are NOT consumed by the reaction
- · Have no effect on the change in Gibbs Free Energy

complex forms. enters active site of enzyme

(2) Enzyme/substrate

enzyme changes shape slightly as substrate binds

Competitive: · Binds to active site Noncompetitive:

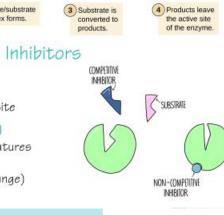
1 Substrate

· Binds to allosteric site

Denaturation

Environmental Temperatures

(outside of optimal range) Salinity



Photosynthesis

Liaht Reactions

Location: Thylakoid Membrane

Startina Material:

Water (electrons)

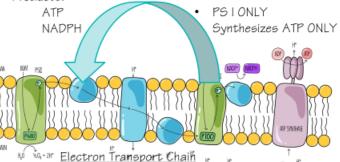
Photons (eneray) Products:

Linear Electron Flow

PSI&PSII

Synthesizes ATP & NADPH

Cyclic Electron Flow



· Protons are pumped into the thylakoid space

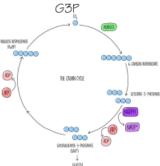
Calvin Cycle

- Location: Stroma
- Starting Material: 3 CO2

9 ATP

6 NADPH

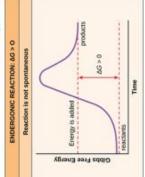
Products:



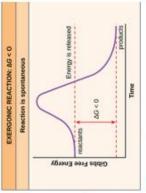
Two Parts: Electron Transport Chain & Chemiosmosis

NADH/FADH, (electrons)

Endergonic Reaction



- Not spontaneous
- ABSORB energy
- Example: ADP + P, → ATP
- Spontaneous RELEASE energy Example: ATP → ADP + P,



Exergonic Reaction

Reactions Energy & ree 599

2

What is Gibbs Free Energy?

Energy available to do work

 $-\Delta G_i$ ΔG_f П ΔG

Change in Gibbs Free Energy

 $T\Delta S$ ∇H II ΔG

Change in Enthalpy

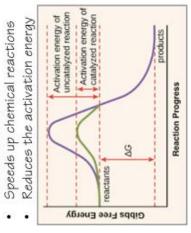
Change in Entropy



Enzymes

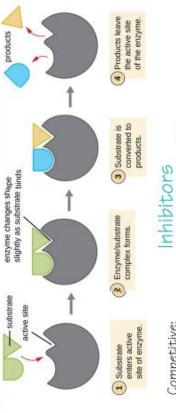
Functions

- Biological catalyst



Important Notes:

- Enzymes are PROTEINS
- Are NOT consumed by the reaction Have no effect on the change in Gibbs Free Energy



Competitive:

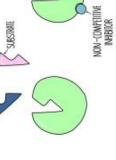
Binds to active site

COMPETITIVE

- Noncompetitive:
- Binds to allosteric site

Environmental Temperatures Denaturation

pH (outside of optimal range) Salinity





Cellular Respiration

Glycolysis

- Location: Cytosol
 - Starting Material:
 - Glucose Products:
- 2 Pyruvate 2 NADH

2 ATP

THE KRIBS OVICE 2× PYRUMITE 2 REDUCED NAD 2 3500019

Oxidative Phosphorylation

- Location: Mitochondrial Cristae
 - Starting Material:
- NADH/FADH₂ (electrons)
 - ATPs Product:
- Two Parts:

Electron Transport Chain & Chemiosmosis

Electron Transport Chain

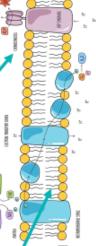
- Protons pumped into IM space
 - Generates proton gradient
- Final electron acceptor: OXYGEN 🤆

Krebs Cycle

- Location: Mitochondrial Matrix
- Staring Material:
- Acetyl CoA
- Products
- 2 CO₂ 3 NADH 1 FADH₂
- 1 ATP

Chemiosmosis

- ATP Synthase uses proton gradient
 - Synthesizes ATP



Photosynthesis

Light Reactions

Location: Thylakoid Membrane

Linear Electron Flow

PS | & PS ||

Water (electrons) Starting Material:

Synthesizes ATP & NADPH

Cyclic Electron Flow

- Photons (energy)
 - Products:

Synthesizes ATP ONLY PS I ONLY NADPH

- Electron Transport Chaiñ
- Protons are pumped into the thylakoid space

- Calvin Cycle
 Location: Stroma
 Starting Material:
 - 3 CO₂
- 9 ATP
- 6 NADPH

